

Oxtoby Chemistry 7th Edition

The Student Solutions manual, authored by Wade Freeman of the University of Illinois at Chicago, contains solutions to the odd numbered problems.

CHEMISTRY FOR ENGINEERING STUDENTS, connects chemistry to engineering, math, and physics; includes problems and applications specific to engineering; and offers realistic worked problems in every chapter that speak to your interests as a future engineer. Packed with built-in study tools, this textbook gives you the resources you need to master the material and succeed in the course. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

The Chemistry Maths Book is a comprehensive textbook of mathematics for undergraduate students of chemistry. Such students often find themselves unprepared and ill-equipped to deal with the mathematical content of their chemistry courses. Textbooks designed to overcome this problem have so far been too basic for complete undergraduate courses and have been unpopular with students. However, this modern textbook provides a complete and up-to-date course companion suitable for all levels of undergraduate chemistry courses. All the most useful and important topics are covered with numerous examples of applications in chemistry and some in physics. The subject is developed in a logical and consistent way with few assumptions of prior knowledge of mathematics. This text is sure to become a widely adopted text and will be highly recommended for all chemistry courses.

How Science Works and its Importance for Science Education

A First Look at Communication Theory

Foye's Principles of Medicinal Chemistry

The Central Science

A textbook of organic chemistry : (for B.Sc. students)

Research in science education has recognized the importance of history and philosophy of science (HPS). Nature of science (NOS) is considered to be an essential part of HPS with important implications for teaching science. The role played by textbooks in developing students ' informed conceptions of NOS has been a source of considerable interest for science educators. In some parts of the world, textbooks become the curriculum and determine to a great extent what is taught and learned in the classroom. Given this background and interest, this monograph has evaluated NOS in university level general chemistry textbooks published in U.S.A. Most textbooks in this study provided little insight with respect to the nine criteria used for evaluating NOS. Some of the textbooks, however, inevitably refer to HPS and thus provide guidelines for future textbooks. A few of the textbooks go into considerable detail to present the atomic models of Dalton, Thomson, Rutherford, Bohr and wave mechanical to illustrate the tentative nature of scientific theories --- an important NOS aspect. These results lead to the question: Are we teaching science as practiced by scientists? An answer to this question can help us to understand the importance of NOS, by providing students an HPS-based environment, so that they too (just like the scientists) feel the thrill and excitement of discovering new things. This monograph provides students and teachers guidelines for introducing various aspects of NOS, based on historical episodes.

This best selling text analyzes the major communication theories at a level appropriate for both lower and upper level courses. The material is presented in a coherent manner and strikes a balance between humanistic, scientific, classical, and contemporary theories. (In contrast to previous editions--no theories have been dropped.) Ethics are covered throughout the text. "A First Look at Communication Theory" is the most widely used textbook for the communication theory course.

This book explores the relationship between the content of chemistry education and the history and philosophy of science (HPS) framework that underlies such education. It discusses the need to present an image that reflects how chemistry developed and progresses. It proposes that chemistry should be taught the way it is practiced by chemists: as a human enterprise, at the interface of scientific practice and HPS. Finally, it sets out to convince teachers to go beyond the traditional classroom practice and explore new teaching strategies. The importance of HPS has been recognized for the science curriculum since the middle of the 20th century. The need for teaching chemistry within a historical context is not difficult to understand as HPS is not far below the surface in any science classroom. A review of the literature shows that the traditional chemistry classroom, curricula, and textbooks while dealing with concepts such as law, theory, model, explanation, hypothesis, observation, evidence and idealization, generally ignore elements of the history and philosophy of science. This book proposes that the conceptual understanding of chemistry requires knowledge and understanding of the history and philosophy of science. " Professor Niaz ' s book is most welcome, coming at a time when there is an urgently felt need to upgrade the teaching of science. The book is a huge aid for adding to the usual way - presenting science as a series of mere facts - also the necessary mandate: to show how science is done, and how science, through its history and philosophy, is part of the cultural development of humanity. " Gerald Holton, Mallinckrodt Professor of Physics & Professor of History of Science, Harvard University " In this stimulating and sophisticated blend of history of chemistry, philosophy of science, and science pedagogy, Professor Mansoor Niaz has succeeded in offering a promising new approach to the teaching of fundamental ideas in chemistry. Historians and philosophers of chemistry --- and above all, chemistry teachers --- will find this book full of valuable and highly usable new ideas " Alan Rocke, Case Western Reserve University " This book artfully connects chemistry and chemistry education to the human context in which chemical science is practiced and the historical and philosophical background that illuminates that practice. Mansoor Niaz deftly weaves together historical episodes in the quest for scientific knowledge with the psychology of learning and philosophical reflections on the nature of scientific knowledge and method. The result is a compelling case for historically and philosophically informed science education. Highly recommended! " Harvey Siegel, University of Miami " Books that analyze the philosophy and history of science in Chemistry are quite rare. ' Chemistry Education and Contributions from History and Philosophy of Science ' by Mansoor Niaz is one of the rare books on the history and philosophy of chemistry and their importance in teaching this science. The book goes through all the main concepts of chemistry, and analyzes the historical and philosophical developments as well as their reflections in textbooks. Closest to my heart is Chapter 6, which is devoted to the chemical bond, the glue that holds together all matter in our earth. The chapter emphasizes the revolutionary impact of the concept of the ' covalent bond ' on the chemical community and the great novelty of the idea that was conceived 11 years before quantum mechanics was able to offer the mechanism of electron pairing and covalent bonding. The author goes then to describe the emergence of two rival theories that explained the nature of the chemical bond in terms of quantum mechanics; these are valence bond (VB) and molecular orbital (MO) theories. He emphasizes the importance of having rival theories and interpretations in science and its advancement. He further argues that this VB-MO rivalry is still alive and together the two conceptual frames serve as the tool kit for thinking and doing chemistry in creative manners. The author surveys chemistry textbooks in the light of the how the books preserve or not the balance between the two theories in describing various chemical phenomena. This Talmudic approach of conceptual tension is a universal characteristic of any branch of evolving wisdom. As such, Mansoor ' s book would be of great utility for chemistry teachers to examine how can they become more effective teachers by recognizing the importance of conceptual tension ". Sason Shaik Saeree K. and Louis P. Fiedler Chair in Chemistry Director, The Lise Meitner-Minerva Center for Computational Quantum Chemistry, The Hebrew University of Jerusalem, ISRAEL

Feyerabend ' s Epistemological Anarchism

The Molecular Science

Chemical Metallurgy

Advanced Chemistry

Organic Chemistry

Prepare for exams and succeed in your chemistry course with this comprehensive solutions manual! Featuring worked out-solutions to the problems in PRINCIPLES OF MODERN CHEMISTRY, 7th Edition, this manual shows you how to approach and solve problems using the same step-by-step explanations found in your textbook examples. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

PRINCIPLES OF MODERN CHEMISTRY has long been considered the standard book for the course, and this modern text has been significantly revised at the sentence level to make it more student-centered and friendly. Authors David W. Oxtoby and H. P. Gillis are now joined by respected researcher and professor, Alan Campion of the University of Texas-Austin, who brings his expertise on surface physics and chemistry and condensed matter spectroscopy to the sixth edition. PRINCIPLES OF MODERN CHEMISTRY has the well-earned reputation of being the most chemically and mathematically accurate and rigorous book on the market, and this edition is no exception. Generated at the Texas Advanced Computing Center at UT-Austin, new mathematically accurate artistic representations of atomic and molecular orbitals will help you easily derive information visually and see how the orbital equations translate into the orbitals' shapes.

Carefully researched by the authors to bring the subject of chemistry up-to-date, this text provides complete coverage of the new A- and AS-level core specifications. The inclusion of objectives and questions make it suitable for self study.

Student Lab Notebook

Calculus

Chapters 1-5

UCLA Chem 20A and Chem 20B

Chemistry

Presents calculus development by integrating technology (with either graphing calculator or computer). The Computational Windows feature offers insights into how technological advances can be used to help understand calculus. Solutions Manual (0-13-178732-2).

NOTE: You are purchasing a standalone product; MasteringA&P does not come packaged with this content. If you would like to purchase both the physical text and MasteringA&P search for ISBN-10: 0321940873/ISBN-13: 9780321940872 . That package includes ISBN-10: 0321943171/ISBN-13: 9780321943170 and ISBN-10: 013389178X/ISBN-13: 9780133891782. " For two-semester general chemistry courses (science majors)."" "Make critical connections in chemistry clear and visibleMcMurry/Fay/Robinson's "Chemistry," Seventh Edition, aims to help students understand the connections between topics in general chemistry and why they matter. The Seventh Edition provides a concise and streamlined narrative that blends the quantitative and visual aspects of chemistry, demonstrates the connections between topics, and illustrates the application of chemistry to their lives and careers. New content offers a better bridge between organic and biochemistry and general chemistry content, and new and improved pedagogical features make the text a true teaching tool rather than just a reference book. New MasteringChemistry features include conceptual worked examples and integrated Inquiry sections that help make critical connections clear and visible and increase students' understanding of chemistry. The Seventh Edition fully integrates the text with new MasteringChemistry content and functionality to support the learning process before, during, and after class. Also Available with MasteringChemistry(R).MasteringChemistry from Pearson is the leading online homework, tutorial, and assessment system, designed to improve results by engaging students before, during, and after class with powerful content. Instructors ensure students arrive ready to learn by assigning educationally effective content before class, and encourage critical thinking and retention with in-class resources such as Learning Catalytics. Students can further master concepts after class through traditional and adaptive homework assignments that provide hints and answer-specific feedback. The Mastering gradebook records scores for all automatically graded assignments in one place, while diagnostic tools give instructors access to rich data to assess student understanding and misconceptions. Mastering brings learning full circle by continuously adapting to each student and making learning more personal than ever-before, during, and after class.

PRINCIPLES OF MODERN CHEMISTRY has dominated the honors and high mainstream general chemistry courses and is considered the standard for the course. The fifth edition is a substantial revision that maintains the rigor of previous editions but reflects the exciting modern developments taking place in chemistry today. Authors David W. Oxtoby and H. P. Gillis provide a unique approach to learning chemical principles that emphasizes the total scientific process—from observation to application—placing general chemistry into a complete perspective for serious-minded science and engineering students. Chemical principles are illustrated by the use of modern materials, comparable to equipment found in the scientific industry. Students are therefore exposed to chemistry and its applications beyond the classroom. This text is perfect for those instructors who are looking for a more advanced general chemistry textbook.

Chemical Principles

Chemistry Insights Ol Twb 2e

Principles and Practice

Principles of Chemistry

An Interdisciplinary Approach to Astrobiology

NOTE: This edition features the same content as the traditional text in a convenient, three-hole-punched, loose-leaf version. Books a la Carte also offer a great value; this format costs significantly less than a new textbook. Before purchasing, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of MyLab(tm)and Mastering(tm) platforms exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a Course ID, provided by your instructor, to register for and use MyLab and Mastering products. For courses in two-semester general chemistry. Accurate, data-driven authorship with expanded interactivity leads to greater student engagement Unrivaled problem sets, notable scientific accuracy and currency, and remarkable clarity have made Chemistry: The Central Science the leading general chemistry text for more than a decade. Trusted, innovative, and calibrated, the text increases conceptual understanding and leads to greater student success in general chemistry by building on the expertise of the dynamic author team of leading researchers and award-winning teachers. In this new edition, the author team draws on the wealth of student data in Mastering(tm)Chemistry to identify where students struggle and strives to perfect the clarity and effectiveness of the text, the art, and the exercises while addressing student misconceptions and encouraging thinking about the practical, real-world use of chemistry. New levels of student interactivity and engagement are made possible through the enhanced eText 2.0 and Mastering Chemistry, providing seamlessly integrated videos and personalized learning throughout the course . Also available with Mastering Chemistry Mastering(tm) Chemistry is the leading online homework, tutorial, and engagement system, designed to improve results by engaging students with vetted content. The enhanced eText 2.0 and Mastering Chemistry work with the book to provide seamless and tightly integrated videos and other rich media and assessment throughout the course. Instructors can assign interactive media before class to engage students and ensure they arrive ready to learn. Students further master concepts through book-specific Mastering Chemistry assignments, which provide hints and answer-specific feedback that build problem-solving skills. With Learning Catalytics(tm) instructors can expand on key concepts and encourage student engagement during lecture through questions answered individually or in pairs and groups. Mastering Chemistry now provides students with the new General Chemistry Primer for remediation of chemistry and math skills needed in the general chemistry course. If you would like to purchase both the loose-leaf version of the text and MyLab and Mastering, search for: 0134557328 / 9780134557328 Chemistry: The Central Science, Books a la Carte Plus MasteringChemistry with Pearson eText -- Access Card Package Package consists of: 0134294165 / 9780134294162 MasteringChemistry with Pearson eText -- ValuePack Access Card -- for Chemistry: The Central Science 0134555635 / 9780134555638 Chemistry: The Central Science, Books a la Carte Edition

This Text Provides A Balanced And Current Treatment Of The Full Spectrum Of Engineering Materials, Covering All The Physical Properties, Applications And Relevant Properties Associated With The Subject. It Explores All The Major Categories Of Materials While Offering Detailed Examinations Of A Wide Range Of New Materials With High-Tech Applications.

Chemical metallurgy is a well founded and fascinating branch of the wide field of metallurgy. This book provides detailed information on both the first steps of separation of desirable minerals and the subsequent mineral processing operations. The complex chemical processes of extracting various elements through hydrometallurgical, pyrometallurgical or electrometallurgical operations are explained. In the choice of material for this work, the author made good use of the synergy of scientific principles and industrial practices, offering the much needed and hitherto unavailable combination of detailed treatises on both compiled in one book.

The Handy Chemistry Answer Book

Chemistry for Engineering Students

Early-onset Neonatal Sepsis

World Religions

Western Traditions

This book argues that the traditional image of Feyerabend is erroneous and that, contrary to common belief, he was a great admirer of science. It shows how Feyerabend presented a vision of science that represented how science really works. Besides giving a theoretical framework based on Feyerabend' s philosophy of science, the book offers criteria that can help readers to evaluate and understand research reported in important international science education journals, with respect to Feyerabend's epistemological anarchism. The book includes an evaluation of general chemistry and physics textbooks. Most science curricula and textbooks provide the following advice to students: Do not allow theories in contradiction with observations, and all scientific theories must be formulated inductively based on experimental facts. Feyerabend questioned this widely prevalent premise of science education in most parts of the world, and in contrast gave the following advice: Scientists can accept a hypothesis despite experimental evidence to the contrary and scientific theories are not always consistent with all the experimental data. No wonder Feyerabend became a controversial philosopher and was considered to be against rationalism and anti-science. Recent research in philosophy of science, however, has shown that most of Feyerabend' s philosophical ideas are in agreement with recent trends in the 21st century. Of the 120 articles from science education journals, evaluated in this book only 9% recognized that Feyerabend was presenting a plurality of perspectives based on how science really works.

Furthermore, it has been shown that Feyerabend could even be considered as a perspectival realist. Among other aspects, Feyerabend emphasized that in order to look for breakthroughs in science one does not have to be complacent about the truth of the theories but rather has to look for opportunities to “break rules” or “violate categories.” Mansoor Niaz carefully analyses references to Feyerabend in the literature and displays the importance of Feyerabend's philosophy in analyzing, historical episodes. Niaz shows through this remarkable

book a deep understanding to the essence of science. – Calvin Kalman, Concordia University, Canada In this book Mansoor Niaz explores the antecedents, context and features of Feyerabend’s work and offers a more-nuanced understanding, then reviews and considers its reception in the science education and philosophy of science literature. This is a valuable contribution to scholarship about Feyerabend, with the potential to inform further research as well as science education practice.– David Geelan, Griffith University, Australia Each topic is treated from the beginning, without assuming prior knowledge. Each chapter starts with an opening section covering an application. These help students to understand the relevance of the topic: they are motivational and they make the text more accessible to the majority of students. Concept Maps have been added, which together with Summaries throughout, aid understanding of main ideas and connections between topics. Margin points highlight key points, making the text more accessible for learning and revision. Checkpoints in each chapter test students’ understanding and support their private study. A selection of questions are included at the end of each chapter, many from past examination papers. Suggested answers are provided in the Answers Key. This innovative, pedagogically driven text explains difficult concepts in a student-oriented manner. The book offers a rigorous and accessible treatment of general chemistry in the context of relevance. Chemistry is presented visually through multi-level images--macroscopic, molecular and symbolic representations--helping students see the connections among the formulas (symbolic), the world around them (macroscopic), and the atoms and molecules that make up the world (molecular). KEY TOPICS: Units of Measurement for Physical and Chemical Change;Atoms and Elements; Molecules, Compounds, and Nomenclature;Chemical Reactions and Stoichiometry;Gases;Thermochemistry;The Quantum-Mechanical Model of the Atom;Periodic Properties of the Elements;Chemical Bonding I: Lewis Theory;Chemical Bonding II: Molecular Shapes, Valence Bond Theory, and Molecular Orbital Theory;Liquids, Solids, and Intermolecular Forces;Solutions;Chemical Kinetics;Chemical Equilibrium;Acids and Bases;Aqueous Ionic Equilibrium;Gibbs Energy and Thermodynamics;Electrochemistry;Radioactivity and Nuclear Chemistry;Organic Chemistry I: Structures;Organic Chemistry II: Reactions;Biochemistry;Chemistry of the Nonmetals;Metals and Metallurgy;Transition Metals and Coordination Compounds MARKET: Appropriate for General Chemistry (2 – Semester) courses.

Principles of Modern Chemistry

The Chemical Evolution of Phosphorus

The Bedford Handbook with 2020 APA Update

Urban Economics

Chemistry Education and Contributions from History and Philosophy of Science

For freshman/sophomore-level courses treating calculus of both one and several variables. Clear and Concise! Varberg focuses on the most critical concepts freeing you to teach the way you want! This popular calculus text remains the shortest mainstream calculus book available - yet covers all the material needed by, and at an appropriate level for, students in engineering, science, and mathematics. It's conciseness and clarity helps students focus on, and understand, critical concepts in calculus without them getting bogged down and lost in excessive and unnecessary detail. It is accurate, without being excessively rigorous, up-to-date without being faddish. The authors make effective use of computing technology, graphics, and applications. Ideal for instructors who want a no-nonsense, concisely written treatment.

Study more effectively and improve your performance at exam time with this comprehensive guide. Written to work hand-in hand with PRINCIPLES OF CHEMISTRY: THE MOLECULAR SCIENCE, 1st Edition, this user-friendly guide includes a wide variety of learning tools to help you master the key concepts of the course.

This comprehensive Fifth Edition has been fully revised and updated to meet the changing curricula of medicinal chemistry courses. The new emphasis is on pharmaceutical care that focuses on the patient, and on the pharmacist a therapeutic clinical consultant, rather than chemist. Approximately 45 contributors, respected in the field of pharmacy education, augment this exhaustive reference. New to this edition are chapters with standardized formats and features, such as Case Studies, Therapeutic Actions, Drug Interactions, and more. Over 700 illustrations supplement this must-have resource.

Student Solutions Manual for Oxtoby, Gillis, and Campion's Principles of Modern Chemistry

Evolving Nature of Objectivity in the History of Science and its Implications for Science Education

Introduction to Materials Science for Engineers

The Central Science (Ap Edition)

The Chemistry Maths Book

Contains discussion, illustrations, and exercises aimed at overcoming common misconceptions; emphasizes on models prevails; and covers topics such as: chemical foundations, types of chemical reactions and solution stoichiometry, electrochemistry, and organic and biological molecules.

Simplifying the complex chemical reactions that take place in everyday through the well-stated answers for more than 600 common chemistry questions, this reference is the go-to guide for students and professionals alike. The book covers everything from the history, major personalities, and groundbreaking reactions and equations in chemistry to laboratory techniques throughout history and the latest developments in the field. Chemistry is an essential aspect of all life that connects with and impacts all branches of science, making this readable resource invaluable across numerous disciplines while remaining accessible at any level of chemistry background. From the quest to make gold and early models of the atom to solar cells, bio-based fuels, and green chemistry and sustainability, chemistry is often at the forefront of technological change and this reference breaks down the essentials into an easily understood format.

This book explores the evolving nature of objectivity in the history of science and its implications for science education. It is generally considered that objectivity, certainty, truth, universality, the scientific method and the accumulation of experimental data characterize both science and science education. Such universal values associated with science may be challenged while studying controversies in their original historical context. The scientific enterprise is not characterized by objectivity or the scientific method, but rather controversies, alternative interpretations of data, ambiguity, and uncertainty. Although objectivity is not synonymous with truth or certainty, it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific. Recent scholarship in history and philosophy of science has shown that it is not the experimental data (Baconian orgy of quantification) but rather the diversity / plurality in a scientific discipline that contributes toward understanding objectivity. History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity. The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others. This book is remarkable for its insistence that the philosophy of science, and in particular that discipline’s analysis of objectivity as the supposed hallmark of the scientific method, is of direct value to teachers of science. Meticulously, yet in a most readable way, Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks; it’s fascinating how certain perspectives fade, while basic questions show no sign of going away. There are few books that take both philosophy and education seriously – this one does! Roald Hoffmann, Cornell University, chemist, writer and Nobel Laureate in Chemistry

Student Solutions Manual for Oxtoby/Gillis' Principles of Modern Chemistry

100 Carbonless Duplicate Sets

A Spectator's Guide to World Religions

Nature of Science in General Chemistry Textbooks

Student Solutions Manual

The world is a very religious place. Wherever you look, people are worshipping, praying, believing, following, even dying for their faith. But are all faiths the same? Do they all call on the same God using different names? Are their beliefs and practices simply cultural expressions of the same spiritual longings? In this timely book, John Dickson presents each of the world's major religions in its best light. He carefully outlines the history, belief systems and spiritual practices of Hinduism, Buddhism, Judaism, Christianity and Islam so that the interested 'spectator' can explore their similarities and differences. For sceptics, believers and students of religion the book provides a fair and friendly introduction to this ultimate subject.

Here is a fascinating reader-friendly exploration of “the phosphorus enigma.” The volume attempts to answer the questions: How did phosphorus atoms, which are produced inside the inner cores of a handful of huge stars, become concentrated in relatively high proportions in the organisms composing Earth’s biosphere? And how did these phosphate derivatives manage to be included in such a great variety of organic molecules playing essential biochemical roles in all known life forms? Due to the interdisciplinary nature of the topic, the volume is arranged in three sections. The first section introduces the fundamental concepts and notions of physics, chemistry, and biology necessary for the proper understanding of the topics discussed within an astronomical framework. The author then focuses on the role of phosphorus and its compounds within the context of chemical evolution in galaxies, considering its relevance in most essential biochemical functions as well as its peculiar chemistry under different physicochemical conditions. The third section provides an overall perspective on the role of phosphorus and its compounds in current areas of research of solid state physics, materials engineering, nanotechnology or medicine.

Early Onset Neonatal Sepsis is covered in this issue of Clinics in Perinatology, guest edited by Drs. Karen Fairchild and Richard Polin. Authorities in the field have come together to pen articles on Innate host defenses and risk for EONS, Group B streptococcus, Diagnosis and management of clinical chorioamnionitis, Molecular diagnostics of sepsis, Use of proteomics in the diagnosis of chorioamnionitis and neonatal sepsis, Adjunct laboratory tests in the diagnosis of EONS, Ureaplasma: role in diseases of prematurity, Meningitis in neonates, Adjunct immunologic therapies in neonatal sepsis, Pathophysiology and treatment of septic shock in neonates, and International perspective on EONS.

A Molecular Approach

A-level Chemistry

An Introduction to the Big Five

If you haven't looked at The Bedford Handbook in a while, look again: This edition has everything your students need to become stronger writers--in a briefer book. This reimagined Bedford Handbook takes a fresh "essentials" approach to the familiar coverage of writing, research, style, and grammar that The Bedford Handbook has always had. The result is a handbook that's equal parts approachable and comprehensive. Students will quickly find answers in the book's direct explanations and step-by-step instruction. They'll get the practice and guidance they need with exercises, how-to guides, model papers, and class-tested examples. The advice you trust from Diana Hacker and Nancy Sommers is here. It all comes in a book that's easier to carry, easier to use, and more affordable than ever.

Now in its fifth edition, this engaging, contributed survey of religions in the Western world - namely Judaism, Christianity, Islam, ancient religions, Indigenous traditions, and new religious movements - explores each tradition in depth from its origins, through its development, to thereligion's meaning and practice in contemporary society.