

Fundamentals Of Chemical Engineering Thermodynamics

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A brand new book, FUNDAMENTALS OF CHEMICAL ENGINEERING THERMODYNAMICS makes the abstract subject of chemical engineering thermodynamics more accessible to undergraduate students. The subject is presented through a problem-solving inductive (from specific to general) learning approach, written in a conversational and approachable manner. Suitable for either a one-semester course or two-semester sequence in the subject, this book covers thermodynamics in a complete and mathematically rigorous manner, with an emphasis on solving practical engineering problems. The approach taken stresses problem-solving, and draws from best practice engineering teaching strategies. FUNDAMENTALS OF CHEMICAL ENGINEERING THERMODYNAMICS uses examples to frame the importance of the material. Each topic begins with a motivational example that is investigated in context to that topic. This framing of the material is helpful to all readers, particularly to global learners who require big picture insights, and hands-on learners who struggle with abstractions. Each worked example is fully annotated with sketches and comments on the thought process behind the solved problems. Common errors are presented and explained. Extensive margin notes add to the book accessibility as well as presenting opportunities for investigation. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

This book explores the use of biomass as an energy source and its application in energy conversion technologies. Focusing on the challenges of, and technologies related to, biomass conversion, the book is divided into three parts. The first part underlines the fundamental concepts that form the basis of biomass production, its feasibility valuation, and its potential utilization. This part does not consider only how biomass is generated, but also methods of assessment. The second part focuses on the clarification of central concepts of the biorefinery processes. After a preliminary introduction with industrial examples, common issues of biochemical reaction engineering applications are analysed in detail. The theory explained in this part demonstrates that the chemical kinetics are the core focus in modelling biological processes such as growth, decay, product formation and feedstock consumption. This part continues with the theory of biofuels production, including biogas, bioethanol, biodiesel and Fischer-Tropsch synthesis of hydrocarbons. The third part of this book gives detailed explanations of preliminary notions related to the theory of thermodynamics. This theory will assist the reader when taking into account the concepts treated in the previous two parts of the book. Several detailed derivations are given to give the reader a full understanding of the arguments at hand. This part also gives literature data on the main properties of some biomass feedstock. Fundamentals of Biofuels Engineering and Technology will be of interest not only to academics and researchers working in this field but also to graduate students and energy professionals seeking to expand their knowledge of this increasingly important area.

This textbook covers the thermodynamics needed by chemical engineers both in their engineering and in their chemistry; it is intended for use in all undergraduate and some graduate-level courses. The authors emphasize a rigorous yet concise presentation of the fundamental chemical concepts governing the behavior of single and multicomponent mixtures, including phase and chemical equilibria. In the application of these concepts, consideration is given to the presentation of experimentally measured thermodynamic properties, and to their prediction for real fluids and their mixtures using methods founded on statistical mechanics. Several applications involving the transfer of heat and work that are of special importance to chemical engineers are studied in detail to show the use of thermodynamics in improving performance. The book is written in SI units and contains worked examples, exercises, and problems.

Chemical Thermodynamics for Process Simulation

Chemical Engineering Thermodynamics II

Small Systems and Fundamentals of Thermodynamics

Generalized Statistical Thermodynamics

Principles of Chemical Engineering Processes

Thermodynamics and information touch every facet of chemistry. However, the physical chemistry curriculum digested by students worldwide is still heavily skewed toward heat/work principles established more than a century ago. Rectifying this situation, Chemical Thermodynamics and Information Theory with Applications explores applications drawn from the intersection of thermodynamics and information theory—two mature and far-reaching fields. In an approach that intertwines information science and chemistry, this book covers: The informational aspects of thermodynamic state equations The algorithmic aspects of transformations—compression, expansion, cyclic, and more The principles of best-practice programming How molecules transmit and modify information via collisions and chemical reactions Using examples from physical and organic chemistry, this book demonstrates how the disciplines of thermodynamics and information theory are intertwined. Accessible to curiosity-driven chemists with knowledge of basic calculus, probability, and statistics, the book provides a fresh perspective on time-honored subjects such as state transformations, heat and work exchanges, and chemical reactions.

This book develops the theory of chemical thermodynamics from first principles, demonstrates its relevance across scientific and engineering disciplines, and shows how thermodynamics can be used as a practical tool for understanding natural phenomena and developing and improving technologies and products. Concepts such as internal energy, enthalpy, entropy, and Gibbs energy are explained using ideas and experiences familiar to students, and realistic examples are given so the usefulness and pervasiveness of thermodynamics becomes apparent. The worked examples illustrate key ideas and demonstrate important types of calculations, and the problems at the end of chapters are designed to reinforce important concepts and show the broad range of applications. Most can be solved using digitized data from open access databases and a spreadsheet. Answers are provided for the numerical problems. A particular theme of the book is the calculation of the equilibrium composition of systems, both reactive and non-reactive, and this includes the principles of Gibbs energy minimization. The overall approach leads to the intelligent use of thermodynamic software packages but, while these are discussed and their use demonstrated, they are not the focus of the book, the aim being to provide the necessary foundations. Another unique aspect is the inclusion of three applications chapters: heat and energy aspects of processing; the thermodynamics of metal production and recycling; and applications of electrochemistry. This book is aimed primarily at students of chemistry, chemical engineering, applied science, materials science, and metallurgy, though it will be also useful for students undertaking courses in geology and environmental science. A solutions

manual is available for instructors.

Applied Chemical Engineering Thermodynamics provides the undergraduate and graduate student of chemical engineering with the basic knowledge, the methodology and the references he needs to apply it in industrial practice. Thus, in addition to the classical topics of the laws of thermodynamics, pure component and mixture thermodynamic properties as well as phase and chemical equilibria the reader will find: - history of thermodynamics - energy conservation - intermolecular forces and molecular thermodynamics - cubic equations of state - statistical mechanics. A great number of calculated problems with solutions and an appendix with numerous tables of numbers of practical importance are extremely helpful for applied calculations. The computer programs on the included disk help the student to become familiar with the typical methods used in industry for volumetric and vapor-liquid equilibria calculations.

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Beyond the Fundamentals

Studyguide for Fundamentals of Chemical Engineering Thermodynamics by Dahm, Kevin D. , ISBN 9781305614161

Thermodynamics with Chemical Engineering Applications

Thermodynamics: Fundamentals and Applications for Chemical Engineers (Second Edition)

Applied Chemical Engineering Thermodynamics

A Practical, Up-to-Date Introduction to Applied Thermodynamics, Including Coverage of Process Simulation Models and an Introduction to Biological Systems Introductory Chemical Engineering Thermodynamics, Second Edition, helps readers master the fundamentals of applied thermodynamics as practiced today: with extensive development of molecular perspectives that enables adaptation to fields including biological systems, environmental applications, and nanotechnology. This text is distinctive in making molecular perspectives accessible at the introductory level and connecting properties with practical implications. Features of the second edition include Hierarchical instruction with increasing levels of detail: Content requiring deeper levels of theory is clearly delineated in separate sections and chapters Early introduction to the overall perspective of composite systems like distillation columns, reactive processes, and biological systems Learning objectives, problem-solving strategies for energy balances and phase equilibria, chapter summaries, and "important equations" for every chapter Extensive practical examples, especially coverage of non-ideal mixtures, which include water contamination via hydrocarbons, polymer blending/recycling, oxygenated fuels, hydrogen bonding, osmotic pressure, electrolyte solutions, zwitterions and biological molecules, and other contemporary issues Supporting software in formats for both MATLAB® and spreadsheets Online supplemental sections and resources including instructor slides, ConcepTests, coursecast videos, and other useful resources

Filling a longstanding gap for graduate courses in the field, Chemical Reaction Engineering: Beyond the Fundamentals covers basic concepts as well as complexities of chemical reaction engineering, including novel techniques for process intensification. The book is divided into three parts: Fundamentals Revisited, Building on Fundamentals, and Beyond

Thermodynamics is the much abused slave of many masters • physicists who love the totally impractical Carnot process, • mechanical engineers who design power stations and refrigerators, • chemists who are successfully synthesizing ammonia and are puzzled by photosynthesis, • meteorologists who calculate cloud bases and predict föhn, boraccia and scirocco, • physico-chemists who vulcanize rubber and build fuel cells, • chemical engineers who rectify natural gas and distill fermented potato juice, • metallurgists who improve steels and harden surfaces, • nutrition counselors who recommend a proper intake of calories, • mechanics who adjust heat exchangers, • architects who construe – and often misconstrue – chneys, • biologists who marvel at the height of trees, • air conditioning engineers who design saunas and the ventilation of air plane cabins, • rocket engineers who create supersonic flows, et cetera. Not all of these professional groups need the full depth and breadth of thermodynamics. For some it is enough to consider a well-stirred tank, for others a stationary nozzle flow is essential, and yet others are well-served with the partial differential equation of heat conduction. It is therefore natural that thermodynamics is prone to mutilation; different group-specific meta-thermodynamics' have emerged which serve the interest of the groups under most circumstances and leave out aspects that are not often needed in their fields.

This book is a very useful reference that contains worked-out solutions for all the exercise problems in the book Chemical Engineering Thermodynamics by the same author. Step-by-step solutions to all exercise problems are provided and solutions are explained with detailed and extensive illustrations. It will come in handy for all teachers and users of Chemical Engineering Thermodynamics.

Fundamentals of Chemical Engineering Thermodynamics, SI Edition

Engineering Thermodynamics

Studyguide for Fundamentals of Chemical Engineering Thermodynamics by Matsoukas, Themis, ISBN 9780132693066

Chemical and Engineering Thermodynamics

Introductory Chemical Engineering Thermodynamics

Thermodynamics: Fundamentals and Applications for Chemical Engineers explores the concepts and properties of thermodynamics and illustrates how they can be applied to solve practical problems. The book introduces the fundamentals of thermodynamics for multi-phase, multi-component systems, providing a framework for dealing with problems in chemical engineering including mixing, compressing, and distilling fluids. The first eight chapters of Thermodynamics focus on single-component thermodynamics, introducing important concepts that will be referenced throughout subsequent chapters. Later chapters introduce modeling for multi-component systems. Topics covered include: properties as a function of state variables; first and second law of thermodynamics; power cycles, combustion, refrigeration cycles, and heat pumps; equilibrium phase relationships; correlations and calculations of vapor-liquid equilibrium data; elementary theories of solutions; and the efficiency of multicomponent separation and reaction processes. The Second Law of Thermodynamics, availability concepts, and process efficiency receive extensive coverage. The clear, well-organized sequence of the chapters helps students successfully learn and retain information. Each of the fifteen chapters includes updated sample problems that underline key principles and problem-solving steps. The book has numerous appendixes for quick reference on everything from conversion factors to Francis constants, and from properties of pure substances to thermodynamics tables and Diagrams. Thermodynamics can be used by chemical, petroleum, and mechanical engineering departments in introductory and intermediate courses on engineering thermodynamics and thermodynamics fundamentals.

A revised edition of the well-received thermodynamics text, this work retains the thorough coverage and excellent organization that made the first edition so popular. Now incorporates

industrially relevant microcomputer programs, with which readers can perform sophisticated thermodynamic calculations, including calculations of the type they will encounter in the lab and in industry. Also provides a unified treatment of phase equilibria. Emphasis is on analysis and prediction of liquid-liquid and vapor-liquid equilibria, solubility of gases and solids in liquids, solubility of liquids and solids in gases and supercritical fluids, freezing point depressions and osmotic equilibria, as well as traditional vapor-liquid and chemical reaction equilibria. Contains many new illustrations and exercises.

Designed as an undergraduate-level textbook in Chemical Engineering, this student-friendly, thoroughly class-room tested book, now in its second edition, continues to provide an in-depth analysis of chemical engineering thermodynamics. The book has been so organized that it gives comprehensive coverage of basic concepts and applications of the laws of thermodynamics in the initial chapters, while the later chapters focus at length on important areas of study falling under the realm of chemical thermodynamics. The reader is thus introduced to a thorough analysis of the fundamental laws of thermodynamics as well as their applications to practical situations. This is followed by a detailed discussion on relationships among thermodynamic properties and an exhaustive treatment on the thermodynamic properties of solutions. The role of phase equilibrium thermodynamics in design, analysis, and operation of chemical separation methods is also deftly dealt with. Finally, the chemical reaction equilibria are skillfully explained. Besides numerous illustrations, the book contains over 200 worked examples, over 400 exercise problems (all with answers) and several objective-type questions, which enable students to gain an in-depth understanding of the concepts and theory discussed. The book will also be a useful text for students pursuing courses in chemical engineering-related branches such as polymer engineering, petroleum engineering, and safety and environmental engineering. New to This Edition • More Example Problems and Exercise Questions in each chapter • Updated section on Vapour-Liquid Equilibrium in Chapter 8 to highlight the significance of equations of state approach • GATE Questions up to 2012 with answers

This Book Is Intended To Present A Good Treatment Of The Fundamentals Of Chemical Engineering Thermodynamics. In This Book Definitions Are Emphasized First To Form The Foundation Of The Subject And Upon This Foundation Arise The First Law, Second Law And The Principle Of Reversibility. Upon This Framework The Secondary Phases Are Based; The Properties Of Real Fluids And Gases, The Concept And Properties Of An Ideal Gas, An Ideal Solution, A Non-Ideal Solution And The Applications Of The Basic Concepts To The Understanding Of The Thermodynamic Aspects Of Compression Processes, Phase Equilibria And Chemical Reaction Equilibria. Sufficient Material Has Been Included To Meet The Requirements Of The Undergraduate Curriculum For A Two-Semester Course In Chemical Engineering Thermodynamics. From A Chemical Engineering Viewpoint, A Significant Emphasis Has To Be Made On The Study And Understanding Of Phase Equilibria And Chemical Reaction Equilibria. These Two Topics Are Covered In Detail In This Book. Illustrations Pertaining To All These Areas/Topics Are Liberally Included Throughout The Text.

Fundamentals of Engineering Thermodynamics

Chemical Thermodynamics and Information Theory with Applications

Chemical Thermodynamics

Fundamental and Advanced Topics

Solutions Manual to Accompany Fundamentals of Engineering Thermodynamics

Building up gradually from first principles, this unique introduction to modern thermodynamics integrates classical, statistical and molecular approaches and is especially designed to support students studying chemical and biochemical engineering. In addition to covering traditional problems in engineering thermodynamics in the context of biology and materials chemistry, students are also introduced to the thermodynamics of DNA, proteins, polymers and surfaces. It includes over 80 detailed worked examples, covering a broad range of scenarios such as fuel cell efficiency, DNA/protein binding, semiconductor manufacturing and polymer foaming, emphasizing the practical real-world applications of thermodynamic principles; more than 300 carefully tailored homework problems, designed to stretch and extend students' understanding of key topics, accompanied by an online solution manual for instructors; and all the necessary mathematical background, plus resources summarizing commonly used symbols, useful equations of state, microscopic balances for open systems, and links to useful online tools and datasets.

This course aims to connect the principles, concepts, and laws/postulates of classical and statistical thermodynamics to applications that require quantitative knowledge of thermodynamic properties from a macroscopic to a molecular level. It covers their basic postulates of classical thermodynamics and their application to transient open and closed systems, criteria of stability and equilibria, as well as constitutive property models of pure materials and mixtures emphasizing molecular-level effects using the formalism of statistical mechanics. Phase and chemical equilibria of multicomponent systems are covered. Applications are emphasized through extensive problem work relating to practical cases.

Principles of Chemical Engineering Processes: Material and Energy Balances introduces the basic principles and calculation techniques used in the field of chemical engineering, providing a solid understanding of the fundamentals of the application of material and energy balances. Packed with illustrative examples and case studies, this book: Discusses problems in material and energy balances related to chemical reactors Explains the concepts of dimensions, units, psychrometry, steam properties, and conservation of mass and energy Demonstrates how MATLAB® and Simulink® can be used to solve complicated problems of material and energy balances Shows how to solve steady-state and transient mass and energy balance problems involving multiple-unit processes and recycle, bypass, and purge streams Develops quantitative problem-solving skills, specifically the ability to think quantitatively (including numbers and units), the ability to translate words into diagrams and mathematical expressions, the ability to use common sense to interpret vague and ambiguous language in problem statements, and the ability to make judicious use of approximations and reasonable assumptions to simplify problems This Second Edition has been updated based upon feedback from professors and students. It features a new chapter related to single- and multiphase

systems and contains additional solved examples and homework problems. Educational software, downloadable exercises, and a solutions manual are available with qualifying course adoption.

Fundamentals of Chemical Engineering ThermodynamicsCengage Learning

Theory and Applications

Solutions Manual For Chemical Engineering Thermodynamics

Introduction to CHEMICAL ENGINEERING THERMODYNAMICS

Molecular Engineering Thermodynamics

Studyguide for Fundamentals of Chemical Engineering Thermodynamics by Dahm, Kevin D., ISBN 9781111580711

Small systems are a very active area of research and development due to improved instrumentation that allows for spatial resolution in the range of sizes from one to 100 nm. In this size range, many physical and chemical properties change, which opens up new approaches to the study of substances and their practical application. This affects both traditional fields of knowledge and many other new fields including physics, chemistry, biology, etc. This book highlights new developments in statistical thermodynamics that answer the most important questions about the specifics of small systems - when one cannot apply equations or traditional thermodynamic models.

The only textbook that applies thermodynamics to real-world process engineering problems This must-read for advanced students and professionals alike is the first book to demonstrate how chemical thermodynamics work in the real world by applying them to actual engineering examples. It also discusses the advantages and disadvantages of the particular models and procedures, and explains the most important models that are applied in process industry. All the topics are illustrated with examples that are closely related to practical process simulation problems. At the end of each chapter, additional calculation examples are given to enable readers to extend their comprehension. Chemical Thermodynamics for Process Simulation instructs on the behavior of fluids for pure fluids, describing the main types of equations of state and their abilities. It discusses the various quantities of interest in process simulation, their correlation, and prediction in detail. Chapters look at the important terms for the description of the thermodynamics of mixtures; the most important models and routes for phase equilibrium calculation; models which are applicable to a wide variety of non-electrolyte systems; membrane processes; polymer thermodynamics; enthalpy of reaction; chemical equilibria, and more. -Explains thermodynamic fundamentals used in process simulation with solved examples -Includes new chapters about modern measurement techniques, retrograde condensation, and simultaneous description of chemical equilibrium -Comprises numerous solved examples, which simplify the understanding of the often complex calculation procedures, and discusses advantages and disadvantages of models and procedures -Includes estimation methods for thermophysical properties and phase equilibria thermodynamics of alternative separation processes -Supplemented with MathCAD-sheets and DDBST programs for readers to reproduce the examples Chemical Thermodynamics for Process Simulation is an ideal resource for those working in the fields of process development, process synthesis, or process optimization, and an excellent book for students in the engineering sciences.

'Chemical engineering is the field of applied science that employs physical, chemical, and biological rate processes for the betterment of humanity'. This opening sentence of Chapter 1 has been the underlying paradigm of chemical engineering. Chemical Engineering: An Introduction is designed to enable the student to explore the activities in which a modern chemical engineer is involved by focusing on mass and energy balances in liquid-phase processes. Problems explored include the design of a feedback level controller, membrane separation, hemodialysis, optimal design of a process with chemical reaction and separation, washout in a bioreactor, kinetic and mass transfer limits in a two-phase reactor, and the use of the membrane reactor to overcome equilibrium limits on conversion. Mathematics is employed as a language at the most elementary level. Professor Morton M. Denn incorporates design meaningfully; the design and analysis problems are realistic in format and scope.

This textbook covers the essential aspects of process safety engineering in a practical and comprehensive manner. It provides readers with an understanding of process safety hazards in the refining and petrochemical industries and how to manage them in a reliable and professional manner. It covers the most important concepts: static electricity, intensity of thermal radiation, thermodynamics of fluid phase equilibria, boiling liquid expanding vapor explosion (BLEVE), emission source models, hazard identification methods, risk control and methods for achieving manufacturing excellence while also focusing on safety. Extensive case studies are included. Aimed at senior undergraduate and graduate chemical engineering students and practicing engineers, this book covers process safety principles and engineering practice authoritatively, with comprehensive examples:

- *Fundamentals, methods, and procedures for the industrial practice of process safety engineering.*
- *The thermodynamic fundamentals and computational methods for release rates from ruptures in pipelines, vessels, and relief valves.*
- *Fundamentals of static electricity hazards and their mitigation.*
- *Quantitative assessment of fires and explosions.*
- *Principles of dispersion calculations for toxic or flammable gases and vapors.*
- *Methods of qualitative and quantitative risk assessment and control.*

Engineering and Chemical Thermodynamics

Chemical Engineering

Thermodynamics for Chemical Engineers

Thermodynamics

The Clear, Well-Organized Introduction to Thermodynamics Theory and Calculations for All Chemical Engineering Undergraduate Students This text is designed to make thermodynamics far easier for undergraduate chemical engineering students to learn, and to help them perform thermodynamic calculations with confidence. Drawing on his award-winning courses at Penn State, Dr. Themis Matsoukas focuses on "why" as well as "how." He offers extensive imagery to help students conceptualize the equations, illuminating thermodynamics with more than 100 figures, as well as 190 examples from within and beyond chemical engineering. Part I clearly introduces the laws of thermodynamics with applications to pure fluids. Part II extends thermodynamics to mixtures, emphasizing phase and chemical equilibrium. Throughout, Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering, including separations, reactions, and capstone design. More than 300 end-of-chapter problems range from basic calculations to realistic environmental applications; these can be solved with any leading mathematical software. Coverage includes • Pure fluids, PVT behavior, and basic calculations of enthalpy and entropy • Fundamental relationships and the calculation of properties from equations of state • Thermodynamic analysis of chemical processes • Phase diagrams of binary and simple ternary systems • Thermodynamics of mixtures using equations of state • Ideal and nonideal solutions • Partial miscibility, solubility of gases and solids, osmotic processes • Reaction equilibrium with applications to single and multiphase reactions The aim of this contemporary textbook is to show students that thermodynamics is a useful tool, not just a series of theoretical exercises. Written in a conversational style, the text presents the second law in a totally new manner--there is no reliance on statistical arguments; instead it is developed as a natural consequence of physical experience. Students are not required to write complex, iterative computer programs to solve phase equilibrium problems--techniques are presented which enable use of readily available math packages. The book also explores electrochemical systems such as batteries and fuel cells. Included in the extensive amount of examples are those which demonstrate the use of thermodynamics in practical design situations.

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Chemical engineers face the challenge of learning the difficult concept and application of entropy and the 2nd Law of Thermodynamics. By following a visual approach and offering qualitative discussions of the role of molecular interactions, Koretsky helps them understand and visualize thermodynamics. Highlighted examples show how the material is applied in the real world. Expanded coverage includes biological content and examples, the Equation of State approach for both liquid and vapor phases in VLE, and the practical side of the 2nd Law. Engineers will then be able to use this resource as the basis for more advanced concepts.

Fundamentals of Process Safety Engineering

Fundamentals of Thermodynamics and Applications

An Introduction to Thermodynamics for Undergraduate Engineering Students

An Introduction

With Historical Annotations and Many Citations from Avogadro to Zermelo

This textbook comprehensively covers the fundamentals and advanced concepts of thermodynamics in a single volume. It provides a detailed discussion of advanced concepts that include energy efficiency, energy sustainability, energy security, organic Rankine cycle, combined cycle power plants, combined cycle power plant integrated with organic Rankine cycle and absorption refrigeration system, integrated coal gasification combined cycle power plants, energy conservation in domestic refrigerators, and next-generation low-global warming potential refrigerants.

Pedagogical features include solved problems and unsolved exercises interspersed throughout the text for better understanding. This textbook is primarily written for senior undergraduate students in the fields of mechanical, automobile, chemical, civil, and aerospace engineering for courses on engineering thermodynamics/thermodynamics and for graduate students in thermal engineering and energy engineering for courses on advanced thermodynamics. It is accompanied by teaching resources, including a solutions manual for instructors. FEATURES Provides design and experimental problems for better understanding Comprehensively discusses power cycles and refrigeration cycles and their advancements Explores the design of energy-efficient buildings to reduce energy consumption Property tables, charts, and multiple-choice questions comprise appendices of the book and are available at <https://www.routledge.com/9780367646288>.

Thermodynamics: Fundamentals and Applications for Chemical Engineers explores the concepts and properties of thermodynamics and illustrates how they can be applied to solve practical problems. The book introduces the fundamentals of thermodynamics for multi-phase, multi-component systems, providing a framework for dealing with problems in chemical engineering

including mixing, compressing, and distilling fluids. The first eight chapters of Thermodynamics focus on single-component thermodynamics, introducing important concepts that will be referenced throughout subsequent chapters. Later chapters introduce modeling for multi-component systems. Topics covered include: properties as a function of state variables; first and second law of thermodynamics; power cycles, combustion, refrigeration cycles, and heat pumps; equilibrium phase relationships; correlations and calculations of vapor-liquid equilibrium data; elementary theories of solutions; and the efficiency of multicomponent separation and reaction processes. The Second Law of Thermodynamics, availability concepts, and process efficiency receive extensive coverage. The clear, well-organized sequence of the chapters helps students successfully learn and retain information. Each of the fifteen chapters includes updated sample problems that underline key principles and problem-solving steps. The book has numerous appendixes for quick reference on everything from conversion factors to Francis constants, and from properties of pure substances to thermodynamics tables and Diagrams. Thermodynamics can be used by chemical, petroleum, and mechanical engineering departments in introductory and intermediate courses on engineering thermodynamics and thermodynamics fundamentals. Born and raised in Chile, Miguel T. Fleischer earned his M.S. and Ph.D. in chemical engineering from the University of Houston where he is an adjunct professor and the undergraduate program director of the Chemical and Biomolecular Engineering Department. Dr. Fleischer worked at Royal Dutch Shell for more than 26 years in research and development, manufacturing, finance, and management. He began teaching when he was an undergraduate student in Chile where he developed a program sponsored by Universidad Catolica de Chile to prepare high school students for college. He was the co-owner and CEO of Fleischer International Trading, a private enterprise that imported and distributed wines from all over the world for 13 years. He continued teaching while he was a graduate student at the University of Houston. He has received the Outstanding Lecturer award of the Cullen College of Engineering four times, the University's Teaching Excellence Award, the Cullen College of Engineering's Career Teaching Award, and the Cullen College of Engineering's Distinguished Engineering Alumni Award.

This book deals with all the concepts in first level Thermodynamics course. Numerous examples are given with the objective of illustrating how the concepts are used for the thermodynamic analysis of devices. Please note: T&F does not sell or distribute the Hardback in India, Pakistan, Nepal, Bhutan, Bangladesh and Sri Lanka

This book, now in its second edition, continues to provide a comprehensive introduction to the principles of chemical engineering thermodynamics and also introduces the student to the application of principles to various practical areas. The book emphasizes the role of the fundamental principles of thermodynamics in the derivation of significant relationships between the various thermodynamic properties. The initial chapter provides an overview of the basic concepts and processes, and discusses the important units and dimensions involved. The ensuing chapters, in a logical presentation, thoroughly cover the first and second laws of thermodynamics, the heat effects, the thermodynamic properties and their relations, refrigeration and liquefaction processes, and the equilibria between phases and in chemical reactions. The book is suitably illustrated with a large number of visuals. In the second edition, new sections on Quasi-Static Process and Entropy Change in Reversible and Irreversible Processes are included. Besides, new Solved Model Question Paper and several new Multiple Choice Questions are also added that help develop the students' ability and confidence in the application of the underlying concepts. Primarily intended for the undergraduate students of chemical engineering and other related engineering disciplines such as polymer, petroleum and pharmaceutical engineering, the book will also be useful for the postgraduate students of the subject as well as professionals in the relevant fields.

A TEXTBOOK OF CHEMICAL ENGINEERING THERMODYNAMICS

Thermodynamics: Fundamentals and Applications for Chemical Engineers

Fundamentals of Chemical Engineering Thermodynamics

With Applications to Chemical Processes

Thermodynamics of Probability Distributions and Stochastic Processes

This book gives the definitive mathematical answer to what thermodynamics really is: a variational calculus applied to probability distributions. Extending Gibbs's notion of ensemble, the Author imagines the ensemble of all possible probability distributions and assigns probabilities to them by selection rules that are fairly general. The calculus of the most probable distribution in the ensemble produces the entire network of mathematical relationships we recognize as thermodynamics. The first part of the book develops the theory for discrete and continuous distributions while the second part applies this thermodynamic calculus to problems in population balance theory and shows how the emergence of a giant component in aggregation, and the shattering transition in fragmentation may be treated as formal phase transitions. While the book is intended as a research monograph, the material is self-contained and the style sufficiently tutorial to be accessible for self-paced study by an advanced graduate student in such fields as physics, chemistry, and engineering.

Master the principles of thermodynamics, and understand their practical real-world applications, with this deep and intuitive undergraduate textbook.

Chemical Reaction Engineering

Chemical Engineering Thermodynamics

Fundamentals of Biofuels Engineering and Technology

Material and Energy Balances, Second Edition