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structure for learning the
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a systematic and logical way. Utilizing clear and consistently colored figures, Principles of Organic Chemistry begins by exploring the step-by-step processes (or mechanisms) by which reactions occur to

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create molecular structures. It then describes some of the many ways these reactions make new compounds, examined by functional groups and corresponding common reaction mechanisms.

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the origin, distribution, chemistry, synthesis, biochemistry, function or use of various classes of naturally occurring substances ranging from small molecules to biopolymers. Each contribution is written by a recognized authority in the field and provides a comprehensive and up-to-date review of the topic in question.

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Addressed to biologists, technologists, and chemists alike, the series can be used by the expert as a source of information and literature citations and by the non-expert as a means of orientation in a rapidly developing discipline.

The authors travel with the reader through the challenging maze of structure

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determination, showing how to distinguish between valuable and deceiving data from IR, NMR and MS spectra, extracting structural conclusions and putting all the pieces together to solve the structure elucidation puzzle. Indeed, human reasoning is key to combining the information contained in those bands,

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signals and peaks by a rationale that enables the makeup of a chemical structure. A number of increasingly more complex problems will act as trip segments and, in addition to the spectra themselves, each chapter is supplemented with figures and tables that decipher the above data and serve as maps for the journey.

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at an unbelievable pace and it becomes more and more difficult to oversee this wealth of data. A crude rationalization of the structures of organic compounds and the atom coordinations can be made with the well-known Kekule model, however, no such generally applicable model exists for the structures of inorganic and

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particularly intermetallic compounds.

There is a need to rationalize the inorganic crystal structures, to find better ways of describing them, of denoting the geometrical relationships between them, of elucidating the electronic factors and of explaining the bonding between the atoms with the aim of not only having a better

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understanding of the known structures, but also of predicting structural features of new compounds.

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have broadened their scopeâ€™ into biology, nanotechnology, materials science, computation, and advanced methods of process systems engineering and controlâ€™ so much that the programs in most chemistry and

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fundamental, molecular-level chemistry to large-scale chemical processing technology. This reflects the way the field has evolved, the synergy at universities between research and education in chemistry and chemical engineering, and the

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way chemists and chemical engineers work together in industry. The astonishing developments in science and engineering during the 20th century have made it possible to dream of new goals that might previously have been considered

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unthinkable. This book identifies the key opportunities and challenges for the chemical sciences, from basic research to societal needs and from terrorism defense to environmental protection, and it looks at the ways in which chemists and chemical

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this relatively new, but increasingly important area of molecular spectroscopy. Key features: A single-source reference on this topic that introduces, describes the background and foundation of this area of spectroscopy. Serves as a

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demonstrate the strong link between chemistry and the various areas of engineering. Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines

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reductions, and health and safety issues related to human exposure. It is an important reference addressing the ongoing concern about the consequences of air

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pollution, and the implementation and modification of the Environmental Protection Agency's (EPA) Clean Air Act. Some of the methods described in the Handbook

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of Air Toxics include
fluorescence, thermal
desorption, selected ion
monitoring, ion
chromatography, light
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subsequent three chapters describe the optical activity, stereoisomerism, biological significance, and industrial applications of these group compounds. Other chapters deal with the reactions and applications

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***Contemporary chemistry is focusing
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molecules and their physicochemical properties. In particular, there has been widespread usage of topological graphs and matrices for the characterization of both individual molecular species and a variety of intermolecular interactions. Our

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prime focus of interest here will center on the distance matrix, and more especially on its derivation for the important class of graphs commonly referred to as chemical trees. The many applications of the distance matrix, $D(G)$, and the Wiener

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branching index, $W(G)$, in chemistry are briefly outlined. $W(G)$ is defined as one half the sum of all the entries in $D(G)$. A recursion formula is developed enabling $W(G)$ to be evaluated for any molecule whose graph G exists in the form of a tree.

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This formula, which represents the first general recursion formula for trees of any kind, is valid irrespective of the valence of the vertices of G or of the degree of branching in G . Several closed expressions giving $W(G)$ for special classes of tree

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molecules are derived from the general formula. One illustrative worked example is also presented. Finally, it is shown how the presence of an arbitrary number of heteroatoms in tree-like molecules can readily be accommodated within our

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general formula by appropriately weighting the vertices and edges of G. Unlike other handbooks in this emerging field, this guide focuses on the challenges and critical parameters in running a metabolomics study, including such often-neglected issues

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as sample preparation, choice of separation and detection method, recording and evaluating data as well as method validation. By systematically covering the entire workflow, from sample preparation to data processing, the insight and

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advice offered here helps to clear the hurdles in setting up and running a successful analysis, resulting in high-quality data from every experiment. Based on more than a decade of practical experience in developing, optimizing and validating

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metabolomics approaches as a routine technology in the academic and industrial research laboratory, the lessons taught here are highly relevant for all systems-level approaches, whether in systems biology, biotechnology, toxicology or

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elucidation both by
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spectroscopic
techniques,
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