

## Chapter 18 Review Chemical Equilibrium

The tools you need to ace your Chemistry II course College success for virtually all science, computing, engineering, and premedical majors depends in part on passing chemistry. The skills learned in chemistry courses are applicable to a number of fields, and chemistry courses are essential to students who are studying to become nurses, doctors, pharmacists, clinical technicians, engineers, and many more among the fastest-growing professions. But if you're like a lot of students who are confused by chemistry, it can seem like a daunting task to tackle the subject. That's where Chemistry II For Dummies can help! Here, you'll get plain-English, easy-to-understand explanations of everything you'll encounter in your Chemistry II class. Whether chemistry is your chosen area of study, a degree requirement, or an elective, you'll get the skills and confidence to score high and enhance your understanding of this often-intimidating subject. So what are you waiting for? Presents straightforward information on complex concepts Tracks to a typical Chemistry II course Serves as an excellent supplement to classroom learning Helps you understand difficult subject matter with confidence and ease Packed with approachable information and plenty of practice

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opportunities, Chemistry II For Dummies is just what you need to make the grade. The Eighth Edition of Zumdahl and DeCoste's best-selling INTRODUCTORY CHEMISTRY: A FOUNDATION that combines enhanced problem-solving structure with substantial pedagogy to enable students to become strong independent problem solvers in the introductory course and beyond. Capturing student interest through early coverage of chemical reactions, accessible explanations and visualizations, and an emphasis on everyday applications, the authors explain chemical concepts by starting with the basics, using symbols or diagrams, and conclude by encouraging students to test their own understanding of the solution. This step-by-step approach has already helped hundreds of thousands of students master chemical concepts and develop problem-solving skills. The book is known for its focus on conceptual learning and for the way it motivates students by connecting chemical principles to real-life experiences in chapter-opening discussions and Chemistry in Focus boxes. The Seventh Edition now adds a questioning pedagogy to in-text examples to help students learn what questions they should be asking themselves while solving problems, offers a revamped art program to better serve visual learners, and includes a significant number of revised end-of-chapter questions. The book's unsurpassed teaching and learning resources include a robust

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technology package that now offers a choice between OWL: Online Web Learning and Enhanced WebAssign. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

In the newly released Eighth Edition of *Chemistry: The Molecular Nature of Matter*, the authors deliver a practical and essential introduction to general chemistry. Thoroughly revised, with particular attention paid to the optimization of the text and included LearnSmart questions, the book focuses throughout on keeping the material accessible and succinct.

This text contains detailed worked solutions to all the end-of-chapter exercises in the textbook *Organic Chemistry*. Notes in tinted boxes in the page margins highlight important principles and comments.

Student Solutions Manual  
Chemistry

Modern Chemistry

The Study of Matter and Its Changes  
Chemistry, Student Study Guide

***Known for its readability and systematic, rigorous approach, this fully updated Ninth Edition of FUNDAMENTALS OF ANALYTICAL CHEMISTRY offers extensive coverage of the principles and practices of analytic chemistry and consistently shows students its applied nature. The book's award-winning authors begin each chapter with a story and photo of how analytic chemistry is applied in industry, medicine, and all the sciences. To further reinforce student***

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*learning, a wealth of dynamic photographs by renowned chemistry photographer Charlie Winters appear as chapter-openers and throughout the text. Incorporating Excel spreadsheets as a problem-solving tool, the Ninth Edition is enhanced by a chapter on Using Spreadsheets in Analytical Chemistry, updated spreadsheet summaries and problems, an Excel Shortcut Keystrokes for the PC insert card, and a supplement by the text authors, EXCEL APPLICATIONS FOR ANALYTICAL CHEMISTRY, which integrates this important aspect of the study of analytical chemistry into the book's already rich pedagogy. New to this edition is OWL, an online homework and assessment tool that includes the Cengage YouBook, a fully customizable and interactive eBook, which enhances conceptual understanding through hands-on integrated multimedia interactivity. Available with InfoTrac Student Collections <http://gocengage.com/infotrac>. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.*

*Check your work and better retain the lessons of Foundations of College Chemistry The Foundations of College Chemistry, 16th Edition: Student Solutions Manual is an essential study resource and aid that allows users of the widely read text to check their work against the detailed solutions contained in this book. This Solutions Manual unlocks the full value of the practice problem- and exercise-focused 16th edition of Foundations of College Chemistry. Students will be able to study efficiently and effectively and learn from common errors with the additional guidance offered by this book.*

**A PERFECT PLAN FOR THE PERFECT SCORE We want**

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*you to succeed on your AP\* exam. That's why we've created this 5-step plan to help you study more effectively, use your preparation time wisely, and get your best score. This easy-to-follow guide offers you a complete review of your AP course, strategies to give you the edge on test day, and plenty of practice with AP-style test questions. You'll sharpen your subject knowledge, strengthen your thinking skills, and build your test-taking confidence with Full-length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules-so you can pick the one that meets your needs The 5-Step Plan helps you get the most out of your study time: Step 1: Set Up Your Study Program Step 2: Determine Your Readiness Step 3: Develop the Strategies Step 4: Review the Knowledge Step 5: Build Your Confidence*

*This fully updated Eighth Edition of CHEMICAL PRINCIPLES provides a unique organization and a rigorous but understandable introduction to chemistry that emphasizes conceptual understanding and the importance of models. Known for helping students develop a qualitative, conceptual foundation that gets them thinking like chemists, this market-leading text is designed for students with solid mathematical preparation. The Eighth Edition features a new section on Solving a Complex Problem that discusses and illustrates how to solve problems in a flexible, creative way based on understanding the fundamental ideas of chemistry and asking and answering key questions. The book is also enhanced by an increase of problem solving techniques in the solutions to the Examples, new student learning aids, new “Chemical Insights” and “Chemistry*

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*Explorers” boxes, and more. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.*

*Chemistry & Chemical Reactivity*

*Using Physical Models of Biomolecules to Teach Concepts of Biochemical Structure in Introductory Undergraduate Chemistry*

*Math Review Toolkit*

*Modern Thermodynamics for Chemists and Biochemists*

*Molecular Dynamics*

Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis. This book reviews the chemistry of inorganic substances as the science of matter, the units of measure used, atoms, atomic structure, thermochemistry, nuclear chemistry, molecules, and ions in action. This text also describes the chemical bonds, the representative elements, the changes of state, water and the hydrosphere (which also covers water pollution and water purification). Water purification occurs in nature through the usual water cycle and by the action of microorganisms. The air flushes dissolved gases and volatile pollutants; when water seeps through the soil, it filters solids as they settle in the bottom of placid lakes.

Microorganisms break down large organic molecules containing mostly carbon, hydrogen, nitrogen, oxygen, sulfur, or phosphorus into harmless molecules and ions. This text notes that natural purification occurs if

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the level of contaminants is not so excessive. This textbook is suitable for both chemistry teachers and students.

Master problem-solving using the detailed solutions in this manual, which contains completely worked-out solutions to all odd end-of-chapter exercises and problems. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

A Perfect Plan for the Perfect Score We want you to succeed on your AP\* exam. That's why we've created this 5-step plan to help you study more effectively, use your preparation time wisely, and get your best score. This easy-to-follow guide offers you a complete review of your AP course, strategies to give you the edge on test day, and plenty of practice with AP-style test questions. You'll sharpen your subject knowledge, strengthen your thinking skills, and build your test-taking confidence with Full-length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules--so you can pick the one that meets your needs The 5-Step Plan helps you get the most out of your study time: Step 1: Set Up Your Study Program Step 2: Determine Your Readiness Step 3: Develop the Strategies Step 4: Review the Knowledge Step 5: Build Your Confidence Topics include: Basics \* Reactions and Periodicity \* Stoichiometry \* Gases \* Thermodynamics \* Spectroscopy, Light, and Electrons \* Bonding \* Solids, Liquids, and Intermolecular Forces \* Solutions and

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Colligative Properties \* Kinetics \* Equilibrium \*  
Electrochemistry \* Nuclear Chemistry \* Organic  
Chemistry \* Experimental

The eleventh edition was carefully reviewed with an eye toward strengthening the content available in OWLv2, end-of-chapter questions, and updating the presentation. Nomenclature changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate learning, and the accompanying Interactive Examples in OWLv2 have been redesigned to better parallel the problem-solving approach in the narrative. New Capstone Problems have been added to a number of chapters. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

With Applications in Chemistry and Chemical Engineering

Chemistry: The Central Science

5 Steps to a 5 AP Chemistry, 2014-2015 Edition

The Principles of Chemical Equilibrium

5 Steps to a 5 AP Chemistry, 2012-2013 Edition

**A Perfect Plan for the Perfect Score We want you to succeed on your AP\* exam. That's why we've created this 5-step plan to help you study more effectively, use your preparation time wisely, and get your best score. This easy-to-follow guide offers you a complete review of your AP course, strategies to give you the edge on test day, and plenty of**

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practice with AP-style test questions. You'll sharpen your subject knowledge, strengthen your thinking skills, and build your test-taking confidence with Full-length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules--so you can pick the one that meets your needs The 5-Step Plan helps you get the most out of your study time: Step 1: Set Up Your Study Program Step 2: Determine Your Readiness Step 3: Develop the Strategies Step 4: Review the Knowledge Step 5: Build Your Confidence Topics include: Reactions and Periodicity, Stoichiometry, Gases, Thermodynamics, Spectroscopy, Light, and Electrons, Bonding, Solids, Liquids, and Intermolecular Forces, Solutions and Colligative Properties, Kinetics, Equilibrium, Electrochemistry, Nuclear Chemistry, and Organic Chemistry Also includes: AP Chemistry practice exams \*AP, Advanced Placement Program, and College Board are registered trademarks of the College Entrance Examination Board, which was not involved in the production of, and does not endorse, this product.

The latest developments in quantum and classical molecular dynamics, related techniques, and their applications to several fields of science and engineering. Molecular simulations include a broad range of methodologies such as Monte Carlo, Brownian dynamics, lattice dynamics, and molecular

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dynamics (MD). Features of this book: • Presents advances in methodologies, introduces quantum methods and lists new techniques for classical MD • Deals with complex systems: biomolecules, aqueous solutions, ice and clathrates, liquid crystals, polymers • Provides chemical reactions, interfaces, catalysis, surface phenomena and solids Although the book is not formally divided into methods and applications, the chapters are arranged starting with those that discuss new algorithms, methods and techniques, followed by several important applications. This new edition of CHEMISTRY continues to incorporate a strong molecular reasoning focus, amplified problem-solving exercises, a wide range of real-life examples and applications, and innovative technological resources. With this text's focus on molecular reasoning, readers will learn to think at the molecular level and make connections between molecular structure and macroscopic properties. The Tenth Edition has been revised throughout and now includes a reorganization of the descriptive chemistry chapters to improve the flow of topics, a new basic math skills Appendix, an updated art program with new talking labels that fully explain what is going on in the figure, and much more. Available with InfoTrac Student Collections <http://goengage.com/infotrac>. Important Notice: Media content referenced within the product description or the product

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text may not be available in the ebook version.

Succeed in chemistry with the clear explanations, problem-solving strategies, and dynamic study tools of CHEMISTRY & CHEMICAL REACTIVITY, 9e. Combining thorough instruction with the powerful multimedia tools you need to develop a deeper understanding of general chemistry concepts, the text emphasizes the visual nature of chemistry, illustrating the close interrelationship of the macroscopic, symbolic, and particulate levels of chemistry. The art program illustrates each of these levels in engaging detail--and is fully integrated with key media components. In addition access to OWLv2 may be purchased separately or at a special price if packaged with this text. OWLv2 is an online homework and tutorial system that helps you maximize your study time and improve your success in the course. OWLv2 includes an interactive eBook, as well as hundreds of guided simulations, animations, and video clips. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Analysing Data, Looking for Patterns and Making Deductions  
Foundations of College Chemistry, Student Solutions Manual  
General Chemistry  
Introductory Chemistry

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Teach your course your way with INTRODUCTORY CHEMISTRY: AN ACTIVE LEARNING APPROACH, 7th Edition. This modular, student-friendly resource allows you to tailor the order of chapters to accommodate your needs, not only by presenting topics so they never assume prior knowledge, but also by including any necessary preview or review information needed to learn that topic. The authors' question-and-answer presentation, which allows students to actively learn chemistry while studying an assignment, is reflected in three words of advice and encouragement repeated throughout the book: Learn It Now! This updated 7th edition leaves no students behind. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. This manual contains answers and detailed solutions to all the in-chapter Exercises, Concept Checks, and Self-Assessment and Review Questions, plus step-by-step solutions to selected odd-numbered end-of-chapter problems. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

This volume contains a selection of the papers presented at the Fourth Symposium on Numerical and Physical Aspects of Aerodynamic Flows, which was held at the California State University, Long Beach, from 16-19 January 1989. It includes the Stewartson Memorial Lecture of Professor J. H. Whitelaw, and is divided into three parts. The first is a collection of papers that describe the status of current technology in two- and three-dimensional steady flows, the second deals with two- and three-dimensional unsteady flows, and the papers in the third address stability and transition. Each of the three parts begins with an overview of current research, as described in the following chapters. The individual papers are edited versions of the selected papers originally submitted to

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the symposium. Four years have passed since the Third Symposium, and certain trends become clear if one compares the papers contained in this volume with those of previous volumes. There are more three- than two-dimensional problems considered in Part 1 and the latter address more difficult problems than in the past, for example, the extension to higher angles of attack, to transonic flow, to leading edge ice accretion, and to thick hydrofoils. The large number of papers in the first part reflects the emphasis of current research and development and the needs of industry. Long considered the standard for honors and high-level mainstream general chemistry courses, PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern, rigorous, and chemically and mathematically accurate text on the market. This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure (Chapter 6), Electrochemistry (Chapter 17), and Molecular Spectroscopy and Photochemistry (Chapter 20). In addition, the text utilizes mathematically accurate and artistic atomic and molecular orbital art, and is student friendly without compromising its rigor. End-of-chapter study aids now focus on only the most important key objectives, equations and concepts, making it easier for students to locate chapter content, while new applications to a wide range of disciplines, such as biology, chemical engineering, biochemistry, and medicine deepen students' understanding of the relevance of chemistry beyond the classroom. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Biomedical Mass Transport and Chemical Reaction  
Chemistry 2e

Problems and Problem Solving in Chemistry Education

Programmed Problem-solving for First-year Chemistry

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From Classical to Quantum Methods

A PERFECT PLAN for the PERFECT SCORE

STEP 1 Set up your study plan with three customized study schedules  
STEP 2 Determine your readiness with an AP-style diagnostic exam  
STEP 3 Develop the strategies that will give you the edge on test day  
STEP 4 Review the terms and concepts you need to score high  
STEP 5 Build your confidence with full-length practice exams

If you think you know the Brown, LeMay Bursten Chemistry text, think again. In response to market request, we have created the third Australian edition of the US bestseller, Chemistry: The Central Science. An extensive revision has taken this text to new heights! Triple checked for scientific accuracy and consistency, this edition is a more seamless and cohesive product, yet retains the clarity, innovative pedagogy, functional problem-solving and visuals of the previous version. All artwork and images are now consistent in quality across the entire text. And with a more traditional and logical organisation of the Organic Chemistry content, this comprehensive

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text is the source of all the information and practice problems students are likely to need for conceptual understanding, development of problem solving skills, reference and test preparation.

### Sample Text

Study more effectively and improve your performance at exam time with this comprehensive guide. The guide includes chapter summaries that highlight the main themes; study goals with section references; lists of important terms; a preliminary test for each chapter that provides an average of 80 drill and concept questions; and answers to the preliminary tests. The Study Guide helps you organize the material and practice applying the concepts of the core text. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

### Section Reviews

### Study Guide for

Whitten/Davis/Peck/Stanley's Chemistry,  
10th

Solutions Manual to Accompany Organic  
Chemistry

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Physicochemical Principles and  
Mathematical Modeling  
With Inorganic Qualitative Analysis

***Reaction Rate Theory and Rare Events bridges the historical gap between these subjects because the increasingly multidisciplinary nature of scientific research often requires an understanding of both reaction rate theory and the theory of other rare events. The book discusses collision theory, transition state theory, RRKM theory, catalysis, diffusion limited kinetics, mean first passage times, Kramers theory, Grote-Hynes theory, transition path theory, non-adiabatic reactions, electron transfer, and topics from reaction network analysis. It is an essential reference for students, professors and scientists who use reaction rate theory or the theory of rare events. In addition, the book discusses transition state search algorithms, tunneling corrections, transmission coefficients, microkinetic models, kinetic Monte Carlo, transition path sampling, and importance sampling methods. The unified treatment in this book explains why chemical reactions and other rare events, while having many common theoretical foundations, often require very different computational modeling strategies. Offers an integrated approach to all simulation theories and reaction network analysis, a unique***

***approach not found elsewhere Gives algorithms in pseudocode for using molecular simulation and computational chemistry methods in studies of rare events Uses graphics and explicit examples to explain concepts Includes problem sets developed and tested in a course range from pen-and-paper theoretical problems, to computational exercises***

***Problem solving is central to the teaching and learning of chemistry at secondary, tertiary and post-tertiary levels of education, opening to students and professional chemists alike a whole new world for analysing data, looking for patterns and making deductions. As an important higher-order thinking skill, problem solving also constitutes a major research field in science education. Relevant education research is an ongoing process, with recent developments occurring not only in the area of quantitative/computational problems, but also in qualitative problem solving. The following situations are considered, some general, others with a focus on specific areas of chemistry: quantitative problems, qualitative reasoning, metacognition and resource activation, deconstructing the problem-solving process, an overview of the working memory hypothesis, reasoning with the electron-pushing formalism, scaffolding organic synthesis skills,***

***spectroscopy for structural characterization in organic chemistry, enzyme kinetics, problem solving in the academic chemistry laboratory, chemistry problem-solving in context, team-based/active learning, technology for molecular representations, IR spectra simulation, and computational quantum chemistry tools. The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry.***

***The 2nd edition of our comprehensive prep guide for the difficult and important MCAT (Medical College Admission Test), with in-depth content reviews, strategies for tackling the exam, and access to 4 full-length practice tests online. Thermodynamics is fundamental to university and college curricula in chemistry, physics, engineering and many life sciences around the world. It is also notoriously difficult for students to understand, learn and apply. What makes this book different, and special, is the clarity of the text. The writing style is fluid, natural and lucid, and everything is explained in a logical and transparent manner. Thermodynamics is a deep, and important, branch of science, and this book does not make it "easy". But it does make it intelligible. This book introduces a new, 'Fourth Law' of Thermodynamics' based on the notion of***

***Gibbs free energy, which underpins almost every application of thermodynamics and which the authors claim is worthy of recognition as a 'law'. The last four chapters bring thermodynamics into the twenty-first century, dealing with bioenergetics (how living systems capture and use free energy), macromolecule assembly (how proteins fold), and macromolecular aggregation (how, for example, virus capsids assemble). This is of great current relevance to students of biochemistry, biochemical engineering and pharmacy, and is covered in very few other texts on thermodynamics. The book also contains many novel and effective examples, such as the explanation of why friction is irreversible, the proof of the depression of the freezing point, and the explanation of the biochemical standard state.***

***Introductory Chemistry: An Active Learning Approach***

***Chemistry II For Dummies***

***5 Steps to a 5 AP Chemistry, 2008-2009 Edition***

***Chemical Principles***

***Principles of Modern Chemistry***

***It has been recognized for more than a thousand years that the function of the brain, like the function of the other organs of the body, is determined by its physical, chemical, and biological properties. Evidence that even its highest functions could be***

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explained by these properties was gathered only in recent years, however; these findings, which clearly have to be confirmed by a great deal of further experimental evidence, indicate that most, if not all, of the functions of the brain are based on its bio chemical and biophysical mechanisms. This at first hearing may sound rather simple, but the ability to understand learning, emotion, perhaps even creativity, on biological terms may well be the most important scientific discovery of all time. Few pieces of knowledge can influence our future health and well-being to the degree that understanding of mental mechanisms will. It has been clearly shown in many ways in the previous volumes of this Handbook that from the biochemical or neurochemical point of view the brain is one of the most active organs. The brain seems stable and in some respects permanent; this is evidence not of inactivity but of carefully controlled homeostasis, of dynamic rather than static equilibrium, with most components undergoing metabolic alterations.

*Introductory Chemistry: An Active Learning Approach*  
Cengage Learning

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon—buckyballs, graphite, and diamond—are

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illustrated at the left, as is the molecule methane,  $\text{CH}_4$ , from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement. Teaches the fundamentals of mass transport with a unique approach emphasizing engineering principles in a biomedical environment Includes a basic review of physiology, chemical thermodynamics, chemical kinetics, mass transport, fluid mechanics and relevant mathematical methods Teaches engineering principles and mathematical modelling useful in the broad range of problems that students will encounter in their academic programs as well as later on in their careers Illustrates principles with examples taken from physiology and medicine or with design problems involving biomedical devices Stresses the simplification of problem formulations based on key geometric and functional features that permit practical analyses of biomedical applications Offers a web site of homework problems associated with each chapter and solutions available to instructors Homework problems related to each

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*chapter are available from a supplementary website (*

*The Princeton Review Mcat*

*Fundamentals of Analytical Chemistry*

*The Molecular Nature of Matter*

*Reaction Rate Theory and Rare Events*

*Key Concept Review Guide for General Chemistry*